

Studies on Development of Sufficiently Chemoselective N-Acylation Reagents: N-Acyl-N-(2,3,4,5,6-pentafluorophenyl)methanesulfonamides

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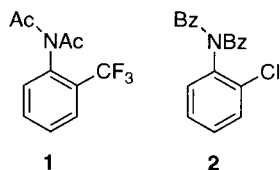
Received 18 May 2000; accepted 12 June 2000

Abstract—A variety of storable *N*-acyl-*N*-(2,3,4,5,6-pentafluorophenyl)methanesulfonamides (**4a–e**) prepared from *N*-2,3,4,5,6-pentafluorophenylmethanesulfonamide (**3**), have been developed after systematic research on the structure–reactivity relationship and were found to serve as *N*-acylation reagents exhibiting sufficiently good chemoselectivity. © 2000 Elsevier Science Ltd. All rights reserved.

Introduction

Chemoselective acylation of amino groups is an important reaction in synthetic operation¹ and is often required in organic synthesis. A variety of reagents have been developed by devising a leaving group for the above purpose,² and continuing efforts have been made in order to develop an ideally chemoselective reagent. It is reasonably assumed that a reagent bearing a bulky leaving group may satisfy such requirement.

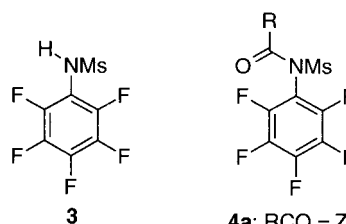
Several years ago, we began studies to develop a new chemoselective *N*-acylation reagent employing the acyl-2-substituted anilide **A** as a leaving group, because the anilide **A** of **1** and **2** is expected to behave as a bulky leaving group (Fig. 1).³ Recently, we have reported that *N*-acetyl-*N*-(2-trifluoromethylphenyl)acetamide^{3a} (**1**) and *N*-benzoyl-*N*-(2-chlorophenyl)benzamide^{3b} (**2**) serve as effective chemoselective *N*-acetylation and *N*-benzoylation reagents, respectively. Calculations of **1** and **2** by using the PM3 method⁴ in the MOPAC package indicated that the 2-substituted phenyl group was placed almost perpendicular to the plane of *N*-C=O (Fig. 2). Therefore, sufficient steric bulkiness would be provided around the acyl carbonyl group.



Keywords: chemoselective *N*-acylation reagent; alkoxyacylation; sulfonamide; amine; protective group.

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As part of our program directed toward development of new *N*-acylation reagents, the present study was undertaken. In this paper, we would like to describe that a variety of storable *N*-acyl-*N*-(2,3,4,5,6-pentafluorophenyl)methanesulfonamides (**4a–e**) were prepared from *N*-2,3,4,5,6-pentafluorophenylmethanesulfonamide (**3**) under mild conditions and were found to serve as *N*-acylation reagents exhibiting sufficiently good chemoselectivity.



- 4a:** RCO = Z
4b: RCO = Alloc (allyloxyacyl)
4c: RCO = Ac
4d: RCO = Bz
4e: RCO = Pv (pivaloyl)

First-Generation Reagents: *N*-Acyl-*N*-(2-fluorophenyl)methanesulfonamide⁵ (**13a**, **13c** and **13d**)

At first, we planned to prepare the *N*-acyl-anilides **6** bearing

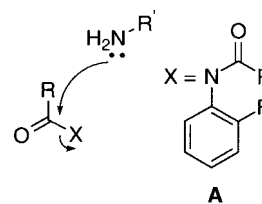


Figure 1. 2-Substituted anilide **A** as a leaving group.

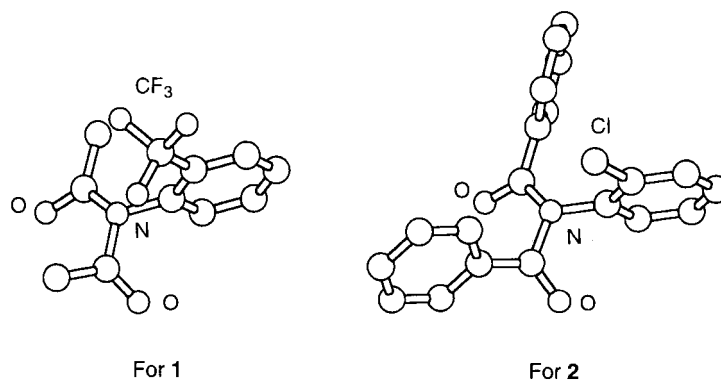
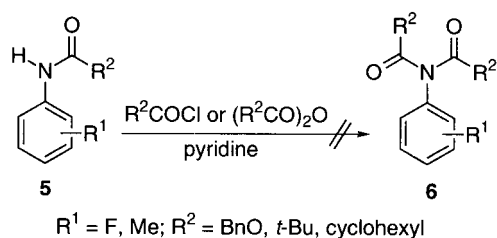


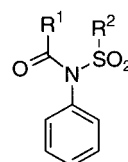
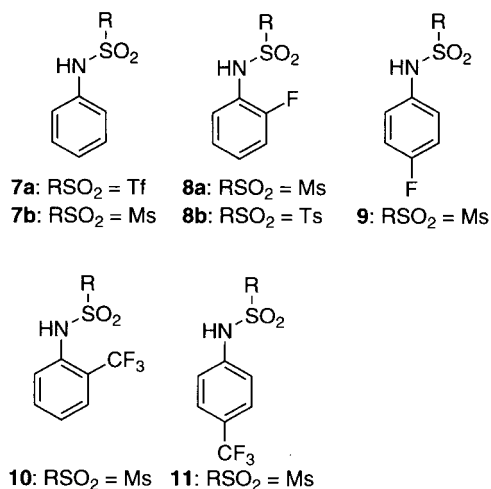
Figure 2. Calculated conformations of **1** and **2**.

other acyl groups except for acetyl and benzoyl groups, under mild reaction conditions, acyl chloride or anhydride in pyridine.⁶ However, after many experiments, our attempts proved to be unsuccessful (Scheme 1).



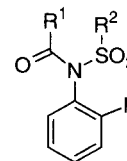
Scheme 1. Many attempts for preparation of various *N*-acylanilides **6**.

In an effort to overcome this problem, we decided to search for a versatile synthon to which various acyl groups could be introduced. We chose to focus on the use of sulfonamides **7–11** as a precursor of various *N*-acylation reagents to circumvent the low reactivity of **5**, since they were expected to have lower pK_a values than **5**. At first, introduction of a *Z* group into sulfonamides **7–11** was examined under mild reaction conditions, 4 equiv. of *Z*Cl in pyridine (Table 1). After several experiments, we found that the *N*-*Z*-sulfonamides **13a**, **13b** and **14–16** could be prepared from the corresponding sulfonamides **8–11** bearing mesyl and tosyl groups in moderate to good yields (entries 3–7). In contrast, the use of trifluoromethanesulfonamide **7a** did not give any acylated product (entry 1).^{7,8}



12a: $R^1CO = Z$, $R^2SO_2 = Tf$

12b: $R^1CO = Z$, $R^2SO_2 = Ms$

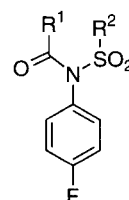


13a: $R^1CO = Z$, $R^2SO_2 = Ms$

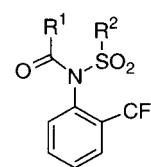
13b: $R^1CO = Z$, $R^2SO_2 = Ts$

13c: $R^1CO = Pv$, $R^2SO_2 = Ms$

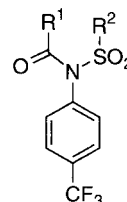
13d: $R^1CO = Bz$, $R^2SO_2 = Ms$



14: $R^1CO = Z$, $R^2SO_2 = Ms$



15: $R^1CO = Z$, $R^2SO_2 = Ms$

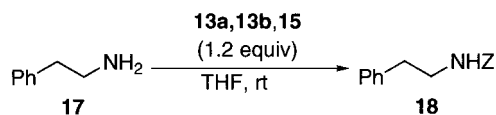


16: $R^1CO = Z$, $R^2SO_2 = Ms$

Table 1. Preparation of *Z*-sulfonamides **12–16** from **7–11** (all reactions were carried out with 4 equiv. of *Z*Cl in pyridine at rt for 1 h)

Entry	Starting sulfonamide	Product (yield %)
1	7a	12a (0), 7a (4)
2	7b	12b (– ^a), 7b (41)
3	8a	13a (90)
4	9	14 (39), 9 (45)
5	10	15 (74)
6	11	16 (21), 11 (64)
7	8b	13b (70)

^a Formation of product **12b** along with significant amounts of inseparable impurities was observed.

Table 2. Acylation potentiality of *N*-Z-sulfonamides **13a**, **13b** and **15**

Entry	Z-Sulfonamide	Time (h)	Yield (%)
1	13a	18	95
2	15	120	80
3	13b	22	97

Table 3. Preparation of *N*-acyl-sulfonamides **13c** and **13d** from **8a** (all reactions were carried out with 2 equiv. of acyl chloride in pyridine at rt for 8 h)

Entry	Acyl chloride	Product: RCO	Yield (%)
1	PvCl	13c :Pv	84
2	BzCl	13d :Bz	92

Since the *N*-Z-sulfonamides **13a**, **13b** and **15** could be prepared in good yields, we chose them as the candidate for an *N*-benzyloxycarbonylation reagent. In order to evaluate their *N*-acyl-transfer potentiality, benzyloxycarbonylation of a typical primary amine, 2-phenylethylamine (**17**), with 1.2 equiv. of **13a**, **13b** and **15** was examined in THF at rt. The reaction was monitored by TLC and the results are shown in Table 2. All of them served as *N*-benzyloxycarbonylation reagents. The *N*-Z-methanesulfonamide **13a** containing a 2-fluoro group on the benzene ring was found to be the most reactive reagent (entry 1). The reactivity of the tosylamide **13b** was somewhat lower than that of **13a** (entry 3). The reaction of the *N*-Z-methanesulfonamide **15** containing a 2-trifluoromethyl group was very sluggish (entry 2). Taking the above results into consideration, **8a** was chosen as a promising precursor for various *N*-acylation reagents.

Introduction of other acyl groups, Pv and Bz groups, into **8a**

was then examined under similar reaction conditions as described above, 2 equiv. of acyl chloride in pyridine (Table 3). As expected, **13c** and **13d** containing a 2-fluoro group could be prepared in good yields.⁹

With **13a**, **13c** and **13d**, acylation of several types of amines was examined in THF. The results are shown in Table 4. Primary, secondary and α -branched amines, **17**, **19** and **20** were all acylated, affording the corresponding acylated products in good yields (entries 1–6). A sterically hindered tertiary alkyl amine **21** was not acylated even in refluxing THF (entries 7 and 8). The substantial difference in the reaction rates between less-hindered and hindered amines prompted us to examine a chemoselective acylation of diamines. As shown in Table 5, less-hindered acylated products were obtained in good yields. However, in all cases, long reaction periods for the *N*-acylation of amines with **13a**, **13c** and **13d** were required.

Second-Generation Reagents: *N*-Acyl-*N*-(2,3,4,5,6-pentafluorophenyl)methanesulfonamides (**4a–e**)

In order to improve the low reactivity of **13a**, **13c** and **13d**, we focused on the preparation of the *N*-sulfonamides **4a** and **25–27** containing a stronger electron-withdrawing group on the benzene ring. Preparation of *N*-acyl-sulfonamides **4a** and **25–27** bearing a Z group was first examined under mild conditions, 4 equiv. of ZCl in pyridine. As shown in Table 6, a variety of storable *N*-Z-sulfonamides **4a**, **25** and **27** could be prepared from the corresponding sulfonamides **3**, **22** and **23** in good yields except **24**. From the viewpoint of synthetic yield, we chose the *N*-acylsulfonamides **4a**, **25** and **26** as the candidate for a second-generation reagent.

In order to evaluate the *N*-acyltransfer potentiality of **4a**, **25** and **26**, benzyloxycarbonylation of **17** was examined with 1.2 equiv. of **4a**, **25** and **26** in THF at rt. The results are shown in Table 7. The *N*-Z-sulfonamides **4a**, **25** and **26** containing 2,3,4,5,6-pentafluoro, 2,6-difluoro and

Table 4. Acylation of several amines with *N*-acyl-sulfonamides **13a**, **13c** and **13d** (all reactions were carried out with 1.2 equiv. of **13a**, **13c** and **13d** in THF)

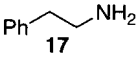
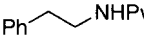
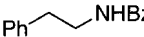
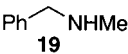
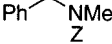
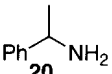
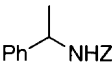
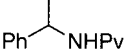
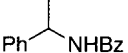
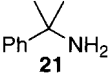
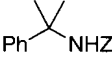
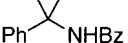
Entry	<i>N</i> -Acyl-sulfonamide	Amine	Conditions	Product	Yield (%)
1	13c		50°C, 20 h		72
2	13d	17	rt, 2 h		91
3	13a		rt, 36 h		93
4	13a		50°C, 20 h		90
5	13c	20	reflux, 24 h		77
6	13d	20	rt, 6 h		94
7	13a		reflux, 24 h		0
8	13d	21	reflux, 24 h		0

Table 5. Chemoselective acylation of diamines with *N*-acyl-sulfonamides **13a** and **13d** (all reactions were carried out with 1.0 equiv. of **13a** or **13d** in THF at 0°C for 24 h and were then left stirring at rt for 12 h)

Entry	13a,d	Diamine	Product and Yield (%)	Entry	13a,d	Diamine	Product and Yield (%)
1	13a		 n=1:82%	3	13a		 91
2	13a	n=2	2:86%	4	13d		 93

Table 6. Preparation of *N*-Z-sulfonamides **4a** and **25–27**

Entry	Starting sulfonamide	Yield (%) ^a
1		4a (84)
2		25 (89)
3		26 (86)
4		27 (32) ^b

^a Yield after recrystallization (benzene–hexane).^b Starting sulfonamide **24** was recovered in 38% yield.

2-fluoro-5-trifluoromethyl groups, acylated the amine **17** in 99, 99 and 96% yields for 1.5, 7 and 7.5 h, respectively.¹⁰

Since **4a** containing 2,3,4,5,6-pentafluoro groups was found to be the most reactive, the introduction of other acyl groups into **3** was examined (Table 8). Fortunately, the storable **4b–e** bearing various acyl groups, Alloc, Ac, Bz, and Pv groups, could be prepared in good yields. We were pleased to find that various acyl groups could be introduced into the sulfonamide **3** under mild conditions.

With **4a–e** bearing various acyl groups in hand, acylation of several types of amines was examined in THF (Table 9). Benzyloxycarbonylation using the *Z*-sulfonamide **4a** gave

Table 7. Acylation potentiality of *N*-Z-sulfonamides **4a**, **25** and **26**

Entry	Sulfonamide	Time (h)	Yield (%)
1	4a	1.5	99
2	25	7	99
3	26	7.5	96

good results. Secondary amine **19** afforded the corresponding *Z*-product in 99% yield for 10 h at rt (entry 5). An α -branched amine **20** afforded the corresponding *Z*-product in 99% yield for 24 h at rt (entry 9). It took 48 h at 65°C for the conversion of a tertiary alkyl amine **21** into the corresponding *Z*-product in 64% yield (entry 14). Acylation using the Alloc- and Ac-sulfonamides **4b** and **4c** gave satisfactory results (entries 1,2,6,7,10,11,15 and 16). The Bz-sulfonamide **4d** was more reactive than the *Z*-, Alloc- and Ac-sulfonamides **4a–c** (for example: entry 12 vs. entries 9–11). Pivaloylation of amines **17** and **20** using **4e** was somewhat less reactive (entries 4 and 13). Thus, the acyl-sulfonamide **4a–e** containing a 2,3,4,5,6-pentafluorophenyl group provided much better reactivity than the first-generation reagents **13a**, **13c** and **13d**. The resulting leaving group **3** of **4a–e** was recovered in >90% yield for recycling by silica gel column.

Since apparent differences in the above reaction rates (Table 9) were observed, selective acylation using 1.0 equiv. of **4a–e** was tested with several diamines as shown in Table 10.¹¹ In all cases, only the less-hindered amino groups were acylated faster in good yields than **13a**, **13b** and **13c** (Tables 4 and 5).

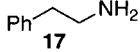
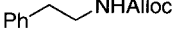



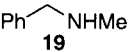
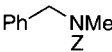
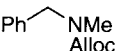
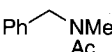

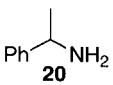
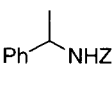
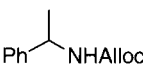
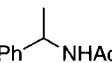
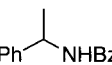
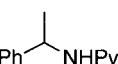
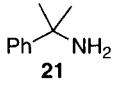
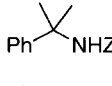
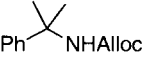
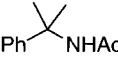
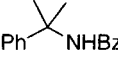
As a result, our new reagents **4a–e** have several advantages as follows: (i) easy preparation of various acylation reagents **4a–e** prepared from a common substrate **3** under mild conditions; (ii) stability in air and easy handling; (iii) good selectivity in acylation. In a molecule with both less-

Table 8. Preparation of *N*-acylsulfonamide **4b–e**

Entry	Conditions	Yield (%)
1	AllocCl (4 equiv.), pyridine, rt, 1 h	4b (91)
2	Ac ₂ O (3 equiv.), pyridine, rt, 4 h	4c (94)
3	BzCl (2 equiv.), pyridine, rt, 8 h	4d (84) ^a
4	PvCl (2 equiv.), pyridine, rt, 8 h	4e (94)

^a Yield after recrystallization (benzene–hexane).

Table 9. Acylation of several amines with *N*-acylsulfonamides **4a–e** (all reactions were carried out with 1.2 equiv. of **4a–e** in THF)

Entry	4a–e : RCO	Amine	Conditions	Product	Yield ^a (%)
1	4b : Alloc		rt, 3.5 h		99
2	4c : Ac	17	rt, 50 min		95
3	4d : Bz	17	rt, 20 min		99
4	4e : Pv	17	rt, 4 h		76 ^b
5	4a : Z		rt, 10 h		99
6	4b : Alloc	19	rt, 24 h		99
7	4c : Ac	19	rt, 60 h		99
8	4d : Bz	19	rt, 1.5 h		99
9	4a : Z		rt, 24 h		99
10	4b : Alloc	20	rt, 24 h		99
11	4c : Ac	20	rt, 4 h		99
12	4d : Bz	20	rt, 1.5 h		99
13	4e : Pv	20	rt, 48 h		62 ^b
14	4a : Z		65 °C, 48 h		64
15	4b : Alloc	21	65 °C, 48 h		68
16	4c : Ac	21	50 °C, 36 h		95
17	4d : Bz	21	65 °C, 48 h		93

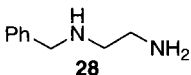
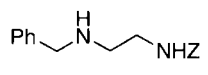
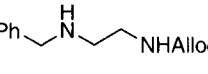
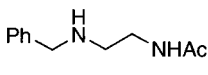
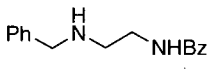
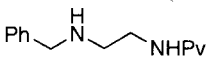
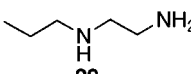
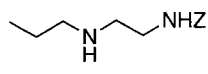
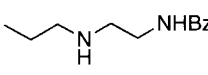
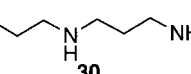
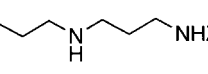
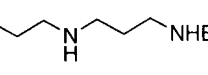
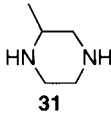
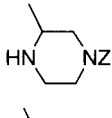
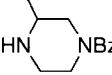
^a Average of two runs.^b ¹H NMR yield.

hindered amino and hindered amino groups, acyltransfer occurred only at the less-hindered amino group.

In summary, a variety of storable *N*-acyl-*N*-(2,3,4,5,6-penta-

fluorophenyl)methanesulfonamides (**4a–e**), which were prepared from 2,3,4,5,6-pentafluorophenylmethanesulfonamide (**3**) under mild conditions, were shown to be *N*-acylation reagents exhibiting sufficiently good chemoselectivity.

Table 10. Chemoselective acylation of several types of amines with *N*-acylsulfonamides **4a–e** (all reactions were carried out with 1.0 equiv. of **4a–e** in THF)

Entry	Amine	N-Acylsulfonamide: RCO	Conditions	Product	Yield ^a (%)
1		4a : Z	0 °C, 24 h		80
2	28	4b : Alloc	0 °C, 24 h		89
3	28	4c : Ac	0 °C, 24 h		92
4	28	4d : Bz	0 °C, 12 h		89
5	28	4e : Pv	0 °C, 36 h		82
6		4a : Z	0 °C, 24 h		73
7	29	4d : Bz	0 °C, 12 h		92
8		4a : Z	0 °C, 24 h		82
9	30	4d : Bz	0 °C, 12 h		91
10		4a : Z	rt, 24 h		85
11	31	4d : Bz	0 °C, 18 h		90

^a Average of two runs.

We believe that these new reagents **4a–e** can be widely used for selective protection of various polyamines.

Experimental

All melting points were determined on a Yanagimoto melting point apparatus and are uncorrected. IR spectra were measured on a JASCO FT/IR-230 diffraction grating infrared spectrophotometer. ¹H and ¹³C NMR spectra were measured on a JEOL GX-400 NMR spectrometer, operating at 400 MHz for ¹H NMR and at 100 MHz for ¹³C NMR. ¹H NMR spectra were reported in δ units, parts per million (ppm) downfield from tetramethylsilane ($\delta=0$) in CDCl₃ and residual DMSO (2.49 ppm) in DMSO-*d*₆. ¹³C NMR spectra were reported in ppm relative to central CDCl₃ reso-

nance ($\delta=77.00$). Electron impact (EI) and FAB-MS were obtained from a JEOL JMS-DX-303 and JMS-HX110 instruments, respectively.

All starting amines and arylamines were commercially available. *N*-Phenylmethanesulfonamide (**7b**), and authentic samples of *N*-(2-phenylethyl)acetamide, *N*-(2-phenylethyl)-pivalamide and *N*-(1-phenylethyl)pivalamide are commercially available. All reagents were available from commercial sources and used without purification. In general, all reactions were carried out in dry solvents under argon atmosphere. The good purity of all isolated products was determined by ¹H NMR, TLC, melting point (for solid compounds) and elemental analysis (for new compounds). All yields (Tables 2, 4, 5 and 7–10) for the acylation of various amines with the *N*-acylsulfonamides

4a–e, 13a–d and **25–27** referred to the average of two isolated yields. Pyridine was distilled under argon atmosphere from CaH_2 before use. THF was distilled under argon atmosphere from sodium/benzophenone ketyl before use. Silica gel column was performed on Kanto Chemical Silica gel 60 (spherical, 100–200 mm) unless otherwise mentioned.

PM3 calculations were performed with CAChe™ MOPAC (version 94.10).

General procedure A for preparation of *N*-arylmethanesulfonamide

To a stirred solution of arylamine in pyridine (0.6 M solution of arylamine) was gradually added methanesulfonyl chloride (1.5 equiv.) at 0°C. The reaction mixture was stirred for 12 h at 23°C and then concentrated directly using a rotary evaporator to afford the crude product as a mixture of *N*-mono- and di-mesylated arylamines. The following work-up was performed to convert the di-mesylated arylamine to mono-mesylated arylamine. The crude product was dissolved in a 2:1 mixture of THF and water (0.4 M solution of the starting arylamine), and then to this mixture was added NaOH (5 equiv. to the starting arylamine). The whole mixture was stirred for 1 h at 23°C, acidified with 15% aqueous HCl and extracted with EtOAc. The organic layer was washed with water and brine, dried over Na_2SO_4 , and concentrated to give the crude *N*-arylmethanesulfonamide.

General procedure B for preparation of *N*-arylmethanesulfonamide

To a stirred solution of arylamine in pyridine (0.6 M solution of arylamine) was gradually added methanesulfonyl chloride (2 equiv.) at 23°C. The reaction mixture was stirred for 6 h at 50°C, allowed to cool and then concentrated directly using a rotary evaporator to afford the crude product as a mixture of *N*-mono- and di-mesylated arylamines. The work-up was performed in the same manner as the procedure A.

***N*-(2-Fluorophenyl)methanesulfonamide (8a).** General procedure A was used with 2-fluoroaniline (1.34 g, 12.1 mmol). Purification by silica gel column (EtOAc–benzene=1:10) followed by recrystallization (EtOAc–hexane) afforded the analytically pure *N*-arylmethanesulfonamide **8a** (1.68 g, 73%) as colorless plates of mp 77–77.5°C. IR (nujol): $\nu=3254, 1612, 1597, 1499, 1466, 1408, 1324, 1254, 1156, 1101 \text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=3.04$ (s, 3H), 6.75 (br s, 1H), 7.08–7.22 (m, 3H), 7.52–7.64 (m, 1H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=39.78, 115.65$ (d, $J=20.0$ Hz), 123.42, 124.49 (d, $J=12.5$ Hz), 124.93 (d, $J=4.2$ Hz), 126.42 (d, $J=7.5$ Hz), 153.90 (d, $J=243.5$ Hz). EI-MS: $m/z=189$ (M^+), 110 (bp), 83. Anal. Calcd for $\text{C}_7\text{H}_8\text{FNO}_2\text{S}$: C, 44.44; H, 4.26; N, 7.40. Found: C, 44.51; H, 4.47; N, 7.44.

***N*-(4-Fluorophenyl)methanesulfonamide (9).** General procedure A was used with 4-fluoroaniline (1.20 mL, 12.7 mmol). Purification by silica gel column (EtOAc–benzene=1:10) followed by recrystallization (EtOAc–hexane)

afforded the analytically pure *N*-arylmethanesulfonamide **9** (1.70 g, 71%) as colorless plates of mp 110–111°C. IR (nujol): $\nu=3243, 1508, 1319, 1146 \text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=2.99$ (s, 3H), 6.84 (br s, 1H), 7.06 (dd, $J=9.0, 9.0$ Hz, 2H), 7.24 (dd, $J=9.0, 4.9$ Hz, 2H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=39.08, 116.29$ (d, $J=22.5$ Hz), 123.66 (d, $J=8.3$ Hz), 132.35 (d, $J=2.5$ Hz), 160.41 (d, $J=245.1$ Hz). EI-MS: $m/z=189$ (M^+), 110, 83 (bp). Anal. Calcd for $\text{C}_7\text{H}_8\text{FNO}_2\text{S}$: C, 44.44; H, 4.26; N, 7.40. Found: C, 44.49; H, 4.55; N, 7.40.

N-(2-Trifluoromethylphenyl)methanesulfonamide (10).

General procedure A was used with 2-trifluoromethylaniline (1.00 g, 6.20 mmol). Purification by silica gel column (EtOAc–benzene=1:5) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-arylmethanesulfonamide **10** (948 mg, 64%) as colorless needles of mp 84°C. IR (nujol): $\nu=3292, 1607, 1494, 1405, 1335, 1318, 1170, 1114 \text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=3.01$ (s, 3H), 6.70 (br s, 1H), 7.33 (dd, $J=8.3, 8.3$ Hz, 1H), 7.60 (dd, $J=8.3, 8.3$ Hz, 1H), 7.66 (d, $J=8.3$ Hz, 1H), 7.83 (d, $J=8.3$ Hz, 1H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=40.08, 121.22$ (q, $J=29.2$ Hz), 123.65 (q, $J=271.8$ Hz), 124.03, 125.43, 126.54 (q, $J=3.8$ Hz), 133.35, 134.10. EI-MS: $m/z=239$ (M^+), 219, 161 (bp), 141, 113. Anal. Calcd for $\text{C}_8\text{H}_8\text{F}_3\text{NO}_2\text{S}$: C, 40.17; H, 3.37; N, 5.86. Found: C, 40.38; H, 3.39; N, 5.87.

N-(4-Trifluoromethylphenyl)methanesulfonamide (11).

General procedure A was used with 4-trifluoromethylaniline (1.13 g, 7.01 mmol). Purification by silica gel column (EtOAc–benzene=1:7) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-arylmethanesulfonamide **11** (1.14 g, 68%) as colorless fine needles of mp 128.5–130°C. IR (nujol): $\nu=3283, 1620, 1330, 1157, 1111 \text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=3.09$ (s, 3H), 7.32 (d, $J=8.5$ Hz, 2H), 7.62 (d, $J=8.5$ Hz, 2H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=40.02, 118.99, 123.73$ (q, $J=271.0$ Hz), 126.83 (q, $J=32.5$ Hz), 126.89 (q, $J=3.3$ Hz), 139.86. EI-MS: $m/z=239$ (M^+), 161, 140, 110, 83 (bp). Anal. Calcd for $\text{C}_8\text{H}_8\text{F}_3\text{NO}_2\text{S}$: C, 40.17; H, 3.37; N, 5.86. Found: C, 40.39; H, 3.56; N, 5.85.

N-(2,3,4,5,6-Pentafluorophenyl)methanesulfonamide (3).

General procedure B was used with 2,3,4,5,6-pentafluoroaniline (2.27 g, 12.4 mmol). Purification by silica gel column (EtOAc–benzene=1:9) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-arylmethanesulfonamide **3** (2.25 g, 69%) as colorless needles of mp 159–161°C. IR (nujol): $\nu=3222, 1530, 1515, 1341, 1330, 1165 \text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=3.25$ (s). The signal of NH was not observed. EI-MS: $m/z=261$ (M^+), 182, 155 (bp), 117, 83. Anal. Calcd for $\text{C}_7\text{H}_4\text{F}_5\text{NO}_2\text{S}$: C, 32.19; H, 1.54; N, 5.36. Found: C, 32.21; H, 1.75; N, 5.36.

N-(2,6-Difluorophenyl)methanesulfonamide (22).

General procedure B was used with 2,6-difluoroaniline (1.45 g, 11.2 mmol). Purification by silica gel column (EtOAc–benzene=1:10) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-arylmethanesulfonamide **22** (1.11 g, 48%) as colorless fine needles of mp 149°C. IR (nujol): $\nu=3251, 1600, 1480, 1335,$

1152 cm⁻¹. ¹H NMR (CDCl₃): δ=3.23 (s, 3H), 6.14 (br s, 1H), 6.94–7.07 (m, 2H), 7.18–7.30 (m, 1H). EI-MS: *m/z*=207 (M⁺), 128 (bp), 149, 101. Anal. Calcd for C₇H₇F₂NO₂S: C, 40.58; H, 3.41; N, 6.76. Found: C, 40.76; H, 3.51; N, 6.72.

***N*-(2-Fluoro-5-trifluoromethylphenyl)methanesulfonamide (23)**. General procedure B was used with 2-fluoro-5-trifluoromethylaniline (1.01 g, 5.64 mmol). Purification by silica gel column (EtOAc–benzene=1:10) followed by recrystallization (EtOAc–hexane) afforded the analytically pure *N*-arylmethanesulfonamide **23** (1.15 g, 79%) as colorless needles of mp 95.5–97.5°C. IR (nujol): ν=3251, 1609, 1523, 1427, 1411, 1335, 1167, 1121 cm⁻¹. ¹H NMR (CDCl₃): δ=3.10 (s, 3H), 6.35–7.10 (br, 1H), 7.27 (dd, *J*=9.0, 9.0 Hz, 1H), 7.45 (ddd, *J*=9.0, 4.6, 2.2 Hz, 1H), 7.88 (dd, *J*=7.3, 2.2 Hz, 1H). EI-MS: *m/z*=257 (M⁺), 178 (bp), 151, 132, 101. Anal. Calcd for C₈H₇F₄NO₂S: C, 37.36; H, 2.74; N, 5.45. Found: C, 37.09; H, 2.83; N, 5.36.

***N*-[3,5-Bis(trifluoromethyl)phenyl]methanesulfonamide (24)**. General procedure B was used with 3,5-bis(trifluoromethyl)aniline (2.40 g, 10.5 mmol). Purification by recrystallization (EtOAc–hexane) afforded the analytically pure *N*-arylmethanesulfonamide **24** (1.84 g, 57%) as colorless needles of mp 149–150°C. IR (nujol): ν=3237, 1624, 1472, 1432, 1375, 1362, 1147, 1124 cm⁻¹. ¹H NMR (CDCl₃): δ=3.14 (s, 3H), 7.31 (br s, 1H), 7.67 (s, 3H). EI-MS: *m/z*=307 (M⁺), 287, 229 (bp), 208, 188, 132, 79. Anal. Calcd for C₉H₇F₆NO₂S: C, 35.19; H, 2.30; N, 4.56. Found: C, 35.24; H, 2.39; N, 4.64.

***N*-2-Fluorophenyl-4-toluenesulfonamide (8b)**. To a stirred solution of 2-fluoroaniline (889 mg, 8.00 mmol) in pyridine (13.0 mL) was gradually added tosyl chloride (3.05 g, 16.0 mmol) at 0°C. The reaction mixture was stirred for 12 h at rt, and then concentrated directly using a rotary evaporator to afford the crude product. Purification by silica gel column (benzene) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-toluenesulfonamide **8b** (1.41 g, 67%) as colorless needles of mp 108–109°C. IR (nujol): ν=3250, 1597, 1499, 1417, 1337, 1255, 1166, 1092 cm⁻¹. ¹H NMR (CDCl₃): δ=2.38 (s, 3H), 6.74 (br s, 1H), 6.93–7.11 (m, 3H), 7.22 (d, *J*=8.5 Hz, 2H), 7.59 (ddd, *J*=8.0, 8.0, 2.2 Hz, 1H), 7.66 (d, *J*=8.5 Hz, 2H). ¹³C NMR (CDCl₃): δ=21.68, 115.27 (d, *J*=20.0 Hz), 123.09, 124.50, 124.58, 124.62, 125.92 (d, *J*=7.5 Hz), 127.05, 129.51, 135.73, 143.97, 153.68 (d, *J*=243.5 Hz). EI-MS: *m/z*=265 (M⁺), 155, 111, 91 (bp). Anal. Calcd for C₁₃H₁₂FNO₂S: C, 58.85; H, 4.56; N, 5.28. Found: C, 59.14; H, 4.70; N, 5.23.

General procedure for benzyloxycarbonylation of *N*-arylsulfonamide

To a stirred solution of *N*-arylmethanesulfonamide in pyridine (0.5 M solution of substrate) was gradually added benzyloxycarbonyl chloride (4 equiv.) at 0°C. The reaction mixture was stirred at 23°C for 1 h and quenched by the addition of water. The whole mixture was extracted with ethyl acetate. The organic extracts were washed with water and brine, dried over Na₂SO₄, and concentrated using a rotary evaporator to afford the crude product.

***N*-Benzyloxycarbonyl-*N*-(2-fluorophenyl)methanesulfonamide (13a)**. General procedure was used with the *N*-arylmethanesulfonamide **8a** (568 mg, 3.00 mmol). Purification by silica gel column (EtOAc–benzene=1:20) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-*Z*-methanesulfonamide **13a** (872 mg, 90%) as colorless needles of mp 97–100°C. IR (nujol): ν=1737, 1497, 1448, 1346, 1155 cm⁻¹. ¹H NMR (CDCl₃): δ=3.42 (s, 3H), 5.21 (s, 2H), 7.12–7.42 (m, 9H). ¹³C NMR (CDCl₃): δ=41.63, 69.63, 116.47, 116.67, 123.32, 123.45, 125.04, 125.08, 128.06, 128.88, 128.92, 131.74, 131.81, 132.54, 134.83, 152.46, 158.67 (d, *J*=250 Hz). EIMS: *m/z*=323 (M⁺), 244, 199, 149, 109, 91 (bp). Anal. Calcd for C₁₅H₁₄FNO₄S: C, 55.72; H, 4.36; N, 4.33. Found: C, 55.87; H, 4.18; N, 4.39.

***N*-Benzyloxycarbonyl-*N*-(4-fluorophenyl)methanesulfonamide (14)**. General procedure was used with the *N*-arylmethanesulfonamide **9** (567 mg, 3.00 mmol). Purification by silica gel column (EtOAc–benzene=1:40) afforded the desired product **14** (451 mg) accompanied by recovery of the unreacted starting material **9** (266 mg, 47%). Further purification of the desired product by recrystallization (benzene–hexane) afforded the analytically pure *N*-*Z*-methanesulfonamide **14** (381 mg, 39%) as colorless fine needles of mp 127–128°C. IR (nujol): ν=1734, 1508, 1351, 1223, 1156 cm⁻¹. ¹H NMR (CDCl₃): δ=3.43 (s, 3H), 5.22 (s, 2H), 7.07–7.14 (m, 2H), 7.18–7.26 (m, 4H), 7.30–7.37 (m, 3H). ¹³C NMR (CDCl₃): δ=41.65, 69.21, 116.26, 116.48, 127.84, 128.10, 130.90, 131.11, 131.21, 134.52, 152.62, 162.86 (d, *J*=250 Hz). EIMS: *m/z*=323 (M⁺), 279, 199, 91 (bp). Anal. Calcd for C₁₅H₁₄FNO₄S: C, 55.72; H, 4.36; N, 4.33. Found: C, 55.43; H, 4.36; N, 4.34.

***N*-Benzyloxycarbonyl-*N*-(2-trifluoromethylphenyl)methanesulfonamide (15)**. General procedure was used with the *N*-arylmethanesulfonamide **10** (749 mg, 3.13 mmol). Purification by silica gel column (EtOAc–benzene=1:20) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-*Z*-methanesulfonamide **15** (865 mg, 74%) as colorless cubes of mp 98–100°C. IR (nujol): ν=1741, 1451, 1360, 1315, 1268, 1167 cm⁻¹. ¹H NMR (CDCl₃): δ=3.50 (s, 3H), 5.16 (d, *J*=12.3 Hz, 1H), 5.23 (d, *J*=12.3 Hz, 1H), 7.13–7.21 (m, 2H), 7.26–7.33 (m, 3H), 7.45 (d, *J*=7.7 Hz, 1H), 7.54 (dd, *J*=7.7, 7.7 Hz, 1H), 7.62 (dd, *J*=7.7, 7.7 Hz, 1H), 7.74 (d, *J*=7.7 Hz, 1H). ¹³C NMR (CDCl₃): δ=41.81, 69.43, 123.11 (q, *J*=274 Hz), 127.27, 127.31, 127.37, 127.42, 127.86, 128.02, 128.08, 128.17, 128.39, 128.54, 128.57, 128.68, 130.05, 132.39, 132.69, 133.02, 134.28, 152.33. EIMS: *m/z*=373 (M⁺), 294, 234, 187, 91 (bp). Anal. Calcd for C₁₆H₁₄F₃NO₄S: C, 51.47; H, 3.78; N, 3.75. Found: C, 51.38; H, 3.76; N, 3.82.

***N*-Benzyloxycarbonyl-*N*-(4-trifluoromethylphenyl)methanesulfonamide (16)**. General procedure was used with the *N*-arylmethanesulfonamide **11** (718 mg, 3.00 mmol). Purification by silica gel column (EtOAc–benzene=1:20) afforded the desired product **16** (423 mg) accompanied by recovery of the unreacted starting material **11** (460 mg, 64%). Further purification of the desired product by recrystallization (benzene–hexane) afforded the analytically pure *N*-*Z*-methanesulfonamide **16** (235 mg, 21%) as colorless needles of mp 148.5–150.5°C. IR (nujol): ν=1746, 1614,

1376, 1335, 1168, 1119 cm^{-1} . ^1H NMR (CDCl_3): $\delta=3.46$ (s, 3H), 5.24 (s, 2H), 7.20–7.25 (m, 2H), 7.32–7.35 (m, 3H), 7.37 (d, $J=8.8$ Hz, 2H), 7.69 (d, $J=8.8$ Hz, 2H). ^{13}C NMR (CDCl_3): $\delta=41.91$, 69.47, 123.54 (q, $J=272$ Hz), 126.43, 126.47, 126.51, 128.01, 128.70, 128.77, 129.96, 131.55 (q, $J=33.1$ Hz), 134.33, 138.12, 152.24. EIMS: $m/z=373$ (M^+), 294, 249, 187, 149, 91, 63 (bp). Anal. Calcd for $\text{C}_{16}\text{H}_{14}\text{F}_3\text{NO}_4\text{S}$: C, 51.47; H, 3.78; N, 3.75, Found: C, 51.49; H, 3.82; N, 3.63.

***N*-Benzyloxycarbonyl-*N*-(2-fluorophenyl)-4-toluenesulfonamide (13b).** General procedure was used with the *N*-aryltoluenesulfonamide **8b** (796 mg, 3.00 mmol). Purification by silica gel column (EtOAc–benzene=1:40) afforded the desired product **13b** (914 mg) accompanied by recovery of the unreacted starting material **8b** (137 mg, 17%). Further purification of the desired product **13b** by recrystallization (benzene–hexane) afforded the analytically pure *N*-*Z*-toluenesulfonamide **13b** (842 mg, 70%) as colorless prisms of mp 103–104°C. IR (nujol): $\nu=1749$, 1519, 1461, 1253, 1363, 1261, 1173 cm^{-1} . ^1H NMR (CDCl_3): $\delta=2.43$ (s, 3H), 5.10 (s, 2H), 7.08–7.30 (m, 9H), 7.36–7.45 (m, 2H), 7.86 (d, $J=8.4$ Hz, 2H). ^{13}C NMR (CDCl_3): $\delta=21.64$, 68.94, 116.28, 116.48, 123.51, 123.65, 124.55, 124.59, 127.75, 128.33, 128.41, 129.08, 129.29, 131.26, 131.33, 131.97, 134.48, 135.67, 145.00, 151.38, 158.70 (d, $J=252$ Hz). EIMS: $m/z=399$ (M^+), 335, 291, 244, 200, 137, 91 (bp), 65. Anal. Calcd for $\text{C}_{21}\text{H}_{18}\text{FNO}_4\text{S}$: C, 63.14; H, 4.54; N, 3.51, Found: C, 63.24; H, 4.56; N, 3.60.

***N*-Benzyloxycarbonyl-*N*-(2,3,4,5,6-pentafluorophenyl)-methanesulfonamide (4a).** General procedure was used with the *N*-arylmethanesulfonamide **3** (742 mg, 2.84 mmol). Purification by silica gel column (EtOAc–benzene=1:20) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-*Z*-methanesulfonamide **4a** (943 mg, 84%) as colorless needles of mp 98–100.5°C. IR (nujol): $\nu=1739$, 1522, 1271, 1174, 993 cm^{-1} . ^1H NMR (CDCl_3): $\delta=3.46$ (s, 3H), 5.29 (s, 2H), 7.27–7.41 (m, 5H). EIMS: $m/z=395$ (M^+), 351, 316, 256, 209, 181, 131, 107, 91, 79, 65 (bp). Anal. Calcd for $\text{C}_{15}\text{H}_{10}\text{F}_5\text{NO}_4\text{S}$: C, 45.58; H, 2.55; N, 3.54, Found: C, 45.65; H, 2.51; N, 3.61.

***N*-Benzyloxycarbonyl-*N*-(2,6-difluorophenyl)methanesulfonamide (25).** General procedure was used with the *N*-arylmethanesulfonamide **22** (624 mg, 3.01 mmol). Purification by silica gel column (EtOAc–benzene=1:20) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-*Z*-methanesulfonamide **25** (914 mg, 89%) as colorless needles of mp 68–69°C. IR (nujol): $\nu=1736$, 1597, 1476, 1362, 1244, 1162 cm^{-1} . ^1H NMR (CDCl_3): $\delta=3.46$ (s, 3H), 5.26 (s, 2H), 7.00 (d, $J=7.1$ Hz, 1H), 7.02 (d, $J=7.1$ Hz, 1H), 7.20–7.43 (m, 6H). ^{13}C NMR (CDCl_3): $\delta=41.58$, 69.59, 111.82, 111.86, 112.06, 112.90, 113.06, 127.71, 128.63, 131.22, 131.32, 131.41, 134.33, 151.43, 159.61 (dd, $J=254$, 3.7 Hz). EIMS: $m/z=341$ (M^+), 262, 217, 202, 155, 127, 107, 91 (bp), 77. Anal. Calcd for $\text{C}_{15}\text{H}_{13}\text{F}_2\text{NO}_4\text{S}$: C, 52.78; H, 3.84; N, 4.10, Found: C, 52.78; H, 3.78; N, 4.09.

***N*-Benzyloxycarbonyl-*N*-(2-fluoro-5-trifluoromethylphenyl)-**

methanesulfonamide (26). General procedure was used with the *N*-arylmethanesulfonamide **23** (824 mg, 3.20 mmol). Purification by silica gel column (EtOAc–benzene=1:20) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-*Z*-methanesulfonamide **26** (1.08 g, 86%) as colorless cubes of mp 89–91°C. IR (nujol): $\nu=1747$, 1514, 1359, 1173, 1126 cm^{-1} . ^1H NMR (CDCl_3): $\delta=3.46$ (s, 3H), 5.26 (s, 2H), 7.21–7.38 (m, 6H), 7.63–7.73 (m, 2H). ^{13}C NMR (CDCl_3): $\delta=41.65$, 69.74, 116.98, 117.18, 122.98 (q, $J=272$ Hz), 123.69, 123.84, 127.91, 128.68, 128.79, 129.78, 129.82, 134.15, 151.58, 160.28 (d, $J=257$ Hz). EIMS: $m/z=391$ (M^+), 312, 292, 252, 205, 177, 91 (bp). Anal. Calcd for $\text{C}_{16}\text{H}_{13}\text{F}_4\text{NO}_4\text{S}$: C, 49.11; H, 3.35; N, 3.58, Found: C, 49.12; H, 3.38; N, 3.62.

***N*-Benzyloxycarbonyl-*N*-[3,5-bis(trifluoromethyl)phenyl]-methanesulfonamide (27).** General procedure was used with the *N*-arylmethanesulfonamide **24** (931 mg, 3.03 mmol). Purification by silica gel column (EtOAc–benzene=1:40) afforded the desired product **27** (622 mg) accompanied by recovery of the unreacted starting material **24** (354 mg, 38%). Further purification of the desired product **27** by recrystallization (benzene–hexane) afforded the analytically pure *N*-*Z*-methanesulfonamide **27** (428 mg, 32%) as colorless fine needles of mp 142–143°C. IR (nujol): $\nu=1743$, 1464, 1374, 1280, 1163, 1125 cm^{-1} . ^1H NMR (CDCl_3): $\delta=3.49$ (s, 3H), 5.25 (s, 2H), 7.20–7.30 (m, 2H), 7.31–7.40 (m, 3H), 7.69 (s, 2H), 7.93 (s, 1H). EIMS: $m/z=441$ (M^+), 422, 362, 302, 236, 188, 91 (bp), 79, 77, 65. Anal. Calcd for $\text{C}_{17}\text{H}_{13}\text{F}_6\text{NO}_4\text{S}$: C, 46.26; H, 2.97; N, 3.17. Found: C, 46.33; H, 3.12; N, 3.27.

***N*-Allyloxycarbonyl-*N*-(2,3,4,5,6-pentafluorophenyl)-methanesulfonamide (4b).** To a stirred solution of the *N*-arylmethanesulfonamide **3** (804 mg, 3.08 mmol) in pyridine (8.0 mL) was gradually added allyloxycarbonyl chloride (1.3 mL, 12.3 mmol) at 0°C. The reaction mixture was stirred at 23°C for 20 min and quenched by the addition of water. The whole mixture was extracted with ethyl acetate. The organic extracts were washed with water and brine, dried over Na_2SO_4 , and concentrated using a rotary evaporator to afford the crude product. Purification by silica gel column (benzene) afforded the analytically pure *N*-Alloc-methanesulfonamide **4b** (968 mg, 91%) as a colorless solid of mp 47–48°C. IR (nujol): $\nu=1749$, 1514, 1365, 1270, 1171 cm^{-1} . ^1H NMR (CDCl_3): $\delta=3.51$ (s, 3H), 4.75 (ddd, $J=5.9$, 1.2, 1.2 Hz, 2H), 5.30 (br dd, $J=1.2$, 1.2 Hz, 1H), 5.33 (ddd, $J=6.6$, 1.2, 1.2 Hz, 1H), 5.88 (ddt, $J=6.6$, 5.9, 1.2 Hz, 1H). EIMS: $m/z=345$ (M^+), 222, 209, 181, 131, 79 (bp). Anal. Calcd for $\text{C}_{11}\text{H}_8\text{F}_5\text{NO}_4\text{S}$: C, 38.27; H, 2.33; N, 4.06, Found: C, 38.46; H, 2.45; N, 4.08.

General procedure for pivaloylation and benzylation of *N*-arylmethanesulfonamide

To a stirred solution of *N*-arylmethanesulfonamide in pyridine (0.5 M solution of substrate) was gradually added acyl chloride (2 equiv.) at 0°C. The reaction mixture was stirred at 23°C for 8 h and quenched by the addition of water. The whole mixture was extracted with EtOAc. The organic extracts were washed with water and brine, dried over Na_2SO_4 , and concentrated using a rotary evaporator to afford the crude product.

***N*-Pivaloyl-*N*-(2-fluorophenyl)methanesulfonamide (13c).** General procedure was used with the *N*-arylmethanesulfonamide **8a** (602 mg, 3.18 mmol). Purification by silica gel column (EtOAc–benzene=1:20) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-Pv-methanesulfonamide **13c** (729 mg, 84%) as colorless cubes of mp 102–103°C. IR (nujol): $\nu=1695, 1494, 1350, 1172, 1116, 968\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.06$ (s, 9H), 3.39 (s, 3H), 7.14–7.27 (m, 2H), 7.42–7.49 (m, 2H). EIMS: $m/z=273$ (M^+), 216, 189 (bp), 172, 151, 137. Anal. Calcd for $\text{C}_{12}\text{H}_{16}\text{FNO}_3\text{S}$: C, 52.73; H, 5.90; N, 5.13, Found: C, 52.57; H, 5.78; N, 5.18.

***N*-Pivaloyl-*N*-phenylmethanesulfonamide (32).** General procedure was used with *N*-phenylmethanesulfonamide (**7b**) (514 mg, 3.00 mmol). Purification by silica gel column (EtOAc–benzene=1:20) afforded the analytically pure *N*-Pv-methanesulfonamide **32** (222 mg, 29%) as colorless solid of mp 121–124°C accompanied by recovery of the unreacted starting material **7b** (293 mg, 57%). IR (nujol): $\nu=1695, 1349, 1119\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.07$ (s, 9H), 3.29 (s, 3H), 7.32–7.36 (m, 2H), 7.42–7.47 (m, 3H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=28.73, 40.22, 43.95, 129.23, 129.69, 130.03, 135.73, 182.16$. EIMS: $m/z=171$ (MH^+-Pv), 149, 119, 92, 85, 57 (bp). Anal. Calcd for $\text{C}_{12}\text{H}_{17}\text{NO}_3\text{S}$: C, 56.45; H, 6.71; N, 5.49, Found: C, 56.16; H, 6.81; N, 5.41.

***N*-Pivaloyl-*N*-(4-fluorophenyl)methanesulfonamide (33).** General procedure was used with the *N*-arylmethanesulfonamide **9** (588 mg, 3.11 mmol). Purification by silica gel column (EtOAc–benzene=1:40) afforded the analytically pure *N*-Pv-methanesulfonamide **33** (204 mg, 24%) as a colorless solid of mp 88–90°C and the unreacted starting material **9** (406 mg, 69%). IR (nujol): $\nu=1693, 1506, 1353, 1218, 1168, 1124\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.08$ (s, 9H), 3.29 (s, 3H), 7.11–7.18 (m, 2H), 7.29–7.36 (m, 2H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=28.72, 40.34, 43.86, 116.37$ (d, $J=23.3$ Hz), 131.64 (d, $J=3.3$ Hz), 131.87 (d, $J=8.3$ Hz), 162.93 (d, $J=251.0$ Hz), 181.78. EIMS: $m/z=273$ (M^+), 189 (bp), 137, 110, 85, 57. Anal. Calcd for $\text{C}_{12}\text{H}_{16}\text{FNO}_3\text{S}$: C, 52.73; H, 5.90; N, 5.13, Found: C, 53.02; H, 5.98; N, 5.00.

***N*-Pivaloyl-*N*-(2,3,4,5,6-pentafluorophenyl)methanesulfonamide (4e).** General procedure was used with the *N*-arylmethanesulfonamide **3** (799 mg, 3.06 mmol). Purification by silica gel column (benzene–hexane=1:1) afforded the analytically pure *N*-Pv-methanesulfonamide **4e** (992 mg, 94%) as a colorless solid. Recrystallization (benzene–hexane) afforded colorless needles of mp 94–95°C. IR (nujol): $\nu=1715, 1706, 1521, 1506, 1362, 1180, 1110, 1036\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.14$ (s, 9H), 3.47 (s, 3H). EIMS: $m/z=345$ (M^+), 261, 209, 183, 131, 85, 57 (bp). Anal. Calcd for $\text{C}_{12}\text{H}_{12}\text{F}_5\text{NO}_3\text{S}$: C, 41.74; H, 3.50; N, 4.06, Found: C, 41.95; H, 3.61; N, 4.14.

***N*-Benzoyl-*N*-(2-fluorophenyl)methanesulfonamide (13d).** General procedure was used with the *N*-arylmethanesulfonamide **8a** (825 mg, 3.16 mmol). Purification by silica gel column (EtOAc–benzene=1:20) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-Bz-methanesulfonamide **13d** (852 mg, 92%) as colorless prisms of mp 105–106°C. IR (nujol): $\nu=1712, 1492, 1351,$

$1247, 1169\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=3.51$ (s, 3H), 6.95–7.02 (m, 1H), 7.13–7.38 (m, 5H), 7.46–7.52 (m, 3H). EIMS: $m/z=293$ (M^+), 185, 139, 105, 77 (bp). Anal. Calcd for $\text{C}_{14}\text{H}_{12}\text{FNO}_3\text{S}$: C, 57.33; H, 4.12; N, 4.78, Found: C, 57.44; H, 3.91; N, 4.80.

***N*-Benzoyl-*N*-(2,3,4,5,6-pentafluorophenyl)methanesulfonamide (4d).** General procedure was used with the *N*-arylmethanesulfonamide **3** (868 mg, 3.32 mmol). Purification by silica gel column (benzene–hexane=1:1) followed by recrystallization (benzene–hexane) afforded the analytically pure *N*-Bz-methanesulfonamide **4d** (1.02 g, 84%) as colorless prisms of mp 105–106°C. IR (nujol): $\nu=1717, 1508, 1447, 1362, 1320, 1260, 1172, 1157\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=3.57$ (s, 3H), 7.32–7.38 (m, 2H), 7.44–7.51 (m, 3H). EIMS: $m/z=365$ (M^+), 238, 105 (bp), 77. Anal. Calcd for $\text{C}_{14}\text{H}_8\text{F}_5\text{NO}_3\text{S}$: C, 46.03; H, 2.21; N, 3.84, Found: C, 46.02; H, 2.34; N, 3.72.

***N*-Acetyl-*N*-(2,3,4,5,6-pentafluorophenyl)methanesulfonamide (4c).** To a stirred solution of the *N*-arylmethanesulfonamide **3** (807 mg, 3.09 mmol) in pyridine (8.0 mL) was gradually added acetic anhydride (0.87 mL, 9.27 mmol) at 0°C. The reaction mixture was stirred at 23°C for 4 h and directly concentrated using a rotary evaporator to afford the crude product. Purification by silica gel column (EtOAc–benzene=1:20) afforded the analytically pure *N*-Ac-methanesulfonamide **4c** (881 mg, 94%) as a colorless oil. IR (neat): $\nu=1728, 1519, 1367, 1237, 1169, 1055\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=2.15$ (s, 3H), 3.49 (s, 3H). EIMS: $m/z=303$ (M^+), 261, 183 (bp), 131, 79. Anal. Calcd for $\text{C}_9\text{H}_6\text{F}_5\text{NO}_3\text{S}$: C, 35.65; H, 2.00; N, 4.62, Found: C, 35.62; H, 2.15; N, 4.65.

Typical procedure for acylation of amine with the *Z*-sulfonamide **4a** corresponding to entry **1** in Table 7

To a stirred solution of 2-phenylethylamine (**17**) (0.11 mL, 0.900 mmol) in THF (3.00 mL) was added the *N*-*Z*-methanesulfonamide **4a** (427 mg, 1.08 mmol) at 23°C. The reaction mixture was stirred for 1.5 h at the same temperature and concentrated using a rotary evaporator. The residue was purified by silica gel[†] column (benzene) to give benzyl *N*-(2-phenylethyl)carbamate (**18**) (227 mg, 99%) as a colorless solid. Successive elution of the silica gel column with EtOAc recovered the sulfonamide **3** (257 mg, 91%).

Benzyl *N*-(2-phenylethyl)carbamate (18). The physical data shown below were comparable to those reported.¹² A colorless solid of mp 66–67.5°C. Reported mp 56–58°C. IR (nujol): $\nu=3326, 1677\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=2.81$ (t, $J=6.8$ Hz, 2H), 3.46 (dt, $J=6.8, 6.8$ Hz, 2H), 4.76 (br s), 5.09 (s, 2H), 7.17–7.38 (m, 10H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=36.17, 42.29, 66.66, 126.37, 127.95, 128.36, 128.48, 128.62, 136.41, 138.51, 156.06$. EI-MS: $m/z=255$ (M^+), 194, 164, 146, 120, 91 (bp).

Physical data of the other acylated amines were shown below.

[†] NH-type silica gel Chromatorex NHDM1020, which was purchased from Fuji Silysia Chemical Ltd, was used.

Benzyl *N*-(1-phenylethyl)carbamate. A colorless oil. IR (neat): $\nu=3328, 1701, 1689\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.48$ (d, $J=6.8$ Hz, 3H), 4.86 (br s, 1H), 4.99–5.10 (m, 1H), 5.05 (d, $J=12.2$ Hz, 1H), 5.11 (d, $J=12.2$ Hz, 1H), 7.20–7.40 (m, 10H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=22.59, 50.78, 66.73, 125.76, 127.17, 127.94, 128.33, 128.48, 136.31, 155.31$. EI-MS: $m/z=255$ (M^+), 240, 196, 163, 106 (bp). Anal. Calcd for $\text{C}_{16}\text{H}_{17}\text{NO}_2$: C, 75.27; H, 6.71; N, 5.49. Found: C, 75.50; H, 6.87; N, 5.55.

Benzyl *N*-benzyl-*N*-methylcarbamate. The physical data shown below were comparable to those reported.^{2d} A colorless oil. IR (neat): $\nu=1704\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=2.87$ and 2.90 (s, 3H), 4.50 (s, 2H), 5.18 (s, 2H), 7.15–7.43 (m, 10H). EI-MS: $m/z=255$ (M^+), 210, 163, 121 (bp), 89.

Benzyl *N*-cumylcarbamate. A colorless solid of mp 63–64°C. IR (nujol): $\nu=3276, 1704\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.67$ (s, 3H \times 2), 5.02 (s, 2H), 5.17 (br s, 1H), 7.21–7.41 (m, 10H). EI-MS: $m/z=269$ (M^+), 254, 210, 119, 108, 91 (bp). Anal. Calcd for $\text{C}_{17}\text{H}_{19}\text{NO}_2$: C, 75.81; H, 7.11; N, 5.20. Found: C, 75.80; H, 7.26; N, 5.31.

Allyl *N*-(2-phenylethyl)carbamate. A colorless oil. IR (neat): $\nu=3338, 1701\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=2.81$ (t, $J=6.8$ Hz, 2H), 3.45 (dt, $J=6.8, 6.8$ Hz, 2H), 4.55 (d, $J=5.4$ Hz, 2H), 4.76 (br s, 1H), 5.20 (d, $J=10.5$ Hz, 1H), 5.28 (d, $J=16.8$ Hz, 1H), 5.90 (ddd, $J=16.8, 10.5, 5.4$ Hz, 1H), 7.13–7.36 (m, 5H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=36.21, 42.24, 65.47, 117.46, 126.35, 128.47, 128.61, 132.77, 138.54, 155.93$. EI-MS: $m/z=205$ (M^+), 164, 114, 105, 91 (bp). Anal. Calcd for $\text{C}_{12}\text{H}_{15}\text{NO}_2$: C, 70.22; H, 7.37; N, 6.83. Found: C, 70.40; H, 7.53; N, 6.94.

Allyl *N*-(1-phenylethyl)carbamate. A colorless solid of mp 43–44°C. IR (nujol): $\nu=3319, 1689\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.48$ (d, $J=6.8$ Hz, 3H), 4.50–4.61 (m, 2H), 4.77–4.91 (br, 1H), 5.01 (br s, 1H), 5.19 (br d, $J=10.3$ Hz, 1H), 5.28 (br d, $J=16.6$ Hz, 1H), 5.82–5.98 (m, 1H), 7.22–7.38 (m, 5H). EI-MS: $m/z=205$ (M^+), 190, 164 (bp), 146, 120, 105, 91. Anal. Calcd for $\text{C}_{12}\text{H}_{15}\text{NO}_2$: C, 70.22; H, 7.37; N, 6.83. Found: C, 70.42; H, 7.54; N, 6.87.

Allyl *N*-benzyl-*N*-methylcarbamate. A colorless oil. IR (neat): $\nu=1702\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=2.86$ and 2.89 (s, 3H), 4.48 (d, $J=10.3$ Hz, 1H), 4.50 (d, $J=10.3$ Hz, 1H), 4.61–4.69 (m, 2H), 5.16–5.37 (m, 2H), 5.88–6.03 (br, 1H), 7.18–7.39 (m, 5H). EI-MS: $m/z=205$ (M^+), 164, 120, 91 (bp). Anal. Calcd for $\text{C}_{12}\text{H}_{15}\text{NO}_2$: C, 70.22; H, 7.37; N, 6.83. Found: C, 70.50; H, 7.43; N, 6.85.

Allyl *N*-cumylcarbamate. A colorless oil. IR (neat): $\nu=3344, 1718, 1707\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.67$ (s, 3H \times 2), 4.48 (d, $J=4.6$ Hz, 2H), 5.12–5.38 (m, 3H), 5.89 (br s, 1H), 7.18–7.47 (m, 5H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=23.87, 55.26, 65.08, 117.39, 124.63, 126.54, 128.19, 132.83, 146.72, 154.02$. EI-MS: $m/z=219$ (M^+ , bp), 178, 161, 146, 142, 117, 98. Anal. Calcd for $\text{C}_{13}\text{H}_{17}\text{NO}_2$: C, 71.21; H, 7.81; N, 6.39. Found: C, 71.45; H, 7.87; N, 6.28.

***N*-(2-Phenylethyl)acetamide.** The physical data shown below were identical with those of the commercially avail-

able *N*-(2-phenylethyl)acetamide and were comparable to those reported.¹³ A colorless solid of mp 53–54°C. Reported mp 50.5–52°C. IR (nujol): $\nu=3287, 1638\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.93$ (s, 3H), 2.82 (t, $J=6.8$ Hz, 2H), 3.51 (dt, $J=6.8, 6.8$ Hz, 2H), 5.58 (br s, 1H), 7.16–7.35 (m, 5H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=23.39, 35.71, 40.71, 126.34, 128.46, 128.54, 138.68, 169.75$. EI-MS: $m/z=163$ (M^+), 104 (bp), 91.

***N*-(1-Phenylethyl)acetamide.** The physical data shown below were comparable to those reported.¹⁴ Colorless needles (benzene–hexane), mp 79–80°C. Reported mp 76°C. IR (KBr): $\nu=3450, 1648\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.49$ (d, $J=6.8$ Hz, 3H), 1.98 (s, 3H), 5.13 (dq, $J=6.8, 6.8$ Hz, 1H), 5.73 (br s, 1H), 7.23–7.40 (m, 5H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=21.82, 23.60, 48.85, 126.05, 127.24, 128.53, 142.96, 168.76$. FAB-MS: $m/z=164$ (MH^+).

***N*-Benzyl-*N*-methylacetamide.** The physical data as shown below were comparable to those reported.¹⁵ A colorless oil. IR (neat): $\nu=1646\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=2.16$ (s, 3H), 2.92 and 2.94 (s, 3H), 4.53 and 4.59 (s, 2H), 7.16–7.40 (m, 5H). EI-MS: $m/z=163$ (M^+ , bp), 148, 120, 106, 91, 77.

***N*-Cumylacetamide.** Colorless needles (benzene–hexane), mp 108–109°C. IR (nujol): $\nu=3337, 3304, 1654\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.69$ (s, 3H \times 2), 1.95 (s, 3H), 5.79 (br s, 1H), 7.19–7.44 (m, 5H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=24.44, 29.16, 55.95, 124.57, 126.48, 128.20, 146.65, 168.80$. EI-MS: $m/z=178$ (MH^+), 177 (M^+), 162, 120 (bp), 91. Anal. Calcd for $\text{C}_{11}\text{H}_{15}\text{NO}$: C, 74.54; H, 8.53; N, 7.90. Found: C, 74.71; H, 8.80; N, 7.97.

***N*-(2-Phenylethyl)benzamide.** The physical data shown below were comparable to those reported.¹³ Colorless needles (EtOAc–hexane), mp 122–123°C. Reported mp 118–120.5°C. IR (nujol): $\nu=3343, 1639\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=2.94$ (t, $J=6.8$ Hz, 2H), 3.73 (dt, $J=6.8, 6.8$ Hz, 2H), 6.11 (br s, 1H), 7.20–7.51 (m, 8H), 7.65–7.72 (m, 2H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=35.81, 41.20, 41.23, 126.44, 126.64, 126.68, 128.39, 128.56, 128.66, 131.22, 134.51, 138.75, 167.20, 167.25$. EI-MS: $m/z=225$ (M^+), 105, 84 (bp).

***N*-(1-Phenylethyl)benzamide.** The physical data shown below were comparable to those reported.¹⁶ Colorless needles (benzene–hexane), mp 128.5–129°C. Reported mp 124°C. IR (nujol): $\nu=3356, 1633\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.61$ (d, $J=7.0$ Hz, 3H), 5.34 (dt, $J=7.0, 7.0$ Hz, 1H), 6.24–6.40 (br, 1H), 7.24–7.51 (m, 8H), 7.74–7.79 (m, 2H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=21.85, 49.27, 126.10, 126.76, 127.31, 128.39, 128.59, 131.29, 134.44, 142.93, 166.29$. FAB-MS: $m/z=226$ (MH^+).

***N*-Benzyl-*N*-methylbenzamide.** The physical data shown below were comparable to those reported.^{3b} A colorless oil. IR (neat): $\nu=1633\text{ cm}^{-1}$. $^1\text{H NMR}$ ($\text{DMSO}-d_6, 90^\circ\text{C}$): $\delta=2.87$ (s, 3H), 4.59 (s, 2H), 7.22–7.48 (m, 10H). FAB-MS: $m/z=226$ (MH^+).

***N*-Cumylbenzamide.** The physical data shown below were comparable to those reported.¹⁷ Colorless needles (benzene–hexane) of mp 168–169°C. Reported mp 156–158°C. IR

(nujol): $\nu=3258, 1635\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.83$ (s, 3H \times 2), 6.39 (br s, 1H), 7.22–7.51 (m, 8H), 7.73–7.78 (m, 2H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=29.26, 56.35, 124.59, 126.64, 126.67, 128.36, 128.39, 131.15, 135.29, 146.62, 166.16$. FAB-MS: $m/z=240$ (MH^+).

Typical procedure for chemoselective acylation of diamine with the *N-Z*-sulfonamide **4a** corresponding to entry **1** in Table 10

To a stirred solution of *N*-benzylethylenediamine (68 μL , 0.450 mmol) in THF (1.5 mL) was added the *N-Z*-sulfonamide **4a** (178 mg, 0.450 mmol) at 0°C . The reaction mixture was stirred for 24 h at the same temperature, quenched by the addition of piperidine (0.05 mL) and directly concentrated using a rotary evaporator. The residue was purified by silica gel column (EtOAc) to give *N*-benzyl-*N'*-benzyloxycarbonylethylenediamine (102 mg, 80%) as a colorless oil together with the sulfonamide **3** (108 mg, 92%).

***N*-Benzyl-*N'*-benzyloxycarbonylethylenediamine.** The physical data shown below were comparable to those reported.^{2w} A colorless oil. IR (neat): $\nu=3332, 1714\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.76$ (br s, 1H), 2.76 (t, $J=5.9$ Hz, 2H), 3.30 (dt, $J=5.9, 5.9$ Hz, 2H), 3.77 (s, 2H), 5.10 (s, 2H), 5.26 (br s, 1H), 7.21–7.39 (m, 10H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=40.82, 48.40, 53.57, 66.64, 126.88, 127.89, 127.93, 128.28, 128.33, 136.46, 139.91, 156.29$. EI-MS: $m/z=285$ (MH^+), 284 (M^+), 194, 149, 133, 132, 118 (bp), 107, 105, 91, 89.

The physical data of the other acylated diamines are shown below.

***N*-Benzyloxycarbonyl-*N'*-propylethylenediamine.** A pale yellow oil. IR (neat): $\nu=3325, 1714\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=0.90$ (t, $J=7.3$ Hz, 3H), 1.47 (tq, $J=7.3, 7.3$ Hz, 2H), 1.56 (br s, 1H), 2.55 (t, $J=7.3$ Hz, 2H), 2.74 (t, $J=5.6$ Hz, 2H), 3.29 (dt, $J=5.6, 5.6$ Hz, 2H), 5.10 (s, 2H), 5.31 (br s, 1H), 7.28–7.39 (m, 5H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=11.80, 23.21, 40.79, 48.83, 51.37, 66.54, 127.83, 127.87, 128.10, 128.27, 136.44, 156.29$. EI-MS: $m/z=237$ (MH^+), 236 (M^+), 207, 149, 91, 72 (bp). Anal. Calcd for $\text{C}_{13}\text{H}_{20}\text{N}_2\text{O}_2$: C, 66.07; H, 8.53; N, 11.85. Found: C, 66.35; H, 8.76; N, 11.60.

***N*-Benzyloxycarbonyl-*N'*-propyl-1,3-propanediamine.** A pale yellow oil. IR (neat): $\nu=3334, 1713, 1701\text{ cm}^{-1}$. $^1\text{H NMR}$ ($\text{CDCl}_3, 50^\circ\text{C}$): $\delta=0.91$ (t, $J=7.1$ Hz, 3H), 1.31 (br s, 1H), 1.48 (tq, $J=7.1, 7.1$ Hz, 2H), 1.66 (tt, $J=6.6, 6.6$ Hz, 2H), 2.54 (t, $J=7.1$ Hz, 2H), 2.68 (t, $J=6.6$ Hz, 2H), 3.25–3.32 (br, 2H), 5.09 (s, 2H), 5.70 (br s, 1H), 7.26–7.40 (m, 5H). EI-MS: $m/z=250$ (M^+), 222, 149 (bp), 134, 120, 91. Anal. Calcd for $\text{C}_{14}\text{H}_{22}\text{N}_2\text{O}_2$: C, 67.16; H, 8.85; N, 11.19. Found: C, 67.34; H, 8.91; N, 10.79.

1-Benzyloxycarbonyl-3-methylpiperazine. A colorless oil. IR (neat): $\nu=3322, 1697\text{ cm}^{-1}$. $^1\text{H NMR}$ ($\text{DMSO}-d_6, 90^\circ\text{C}$): $\delta=0.95$ (d, $J=6.2$ Hz, 3H), 2.39–2.63 (m, 4H), 2.75–2.83 (m, 2H), 3.77–3.83 (m, 2H), 5.08 (s, 2H), 7.28–7.38 (m, 5H). EI-MS: $m/z=234$ (M^+), 233, 134, 91 (bp), 78. Anal. Calcd for $\text{C}_{13}\text{H}_{17}\text{N}_2\text{O}_2$: C, 66.93; H, 7.34; N, 12.01. Found: C, 67.11; H, 7.76; N, 11.97.

***N*-Allyloxycarbonyl-*N'*-benzylethylenediamine.** A pale yellow oil. IR (neat): $\nu=3331, 1714\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.51$ (s, 1H), 2.77 (t, $J=5.9$ Hz, 2H), 3.29 (dt, $J=5.9, 5.9$ Hz, 2H), 3.78 (s, 2H), 4.56 (br d, $J=5.4$ Hz, 2H), 5.16–5.27 (br, 1H), 5.20 (ddt, $J=10.5, 1.2, 1.2$ Hz, 1H), 5.92 (ddt, $J=17.3, 1.2, 1.2$ Hz, 1H), 5.92 (ddt, $J=17.3, 10.5, 5.4$ Hz, 1H), 7.22–7.36 (m, 5H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=40.78, 48.42, 53.58, 65.50, 117.44, 126.90, 127.90, 128.29, 132.81, 139.90, 156.17$. EI-MS: $m/z=235$ (MH^+), 234 (M^+), 204, 164, 143, 120, 106, 91 (bp), 65. Anal. Calcd for $\text{C}_{13}\text{H}_{18}\text{N}_2\text{O}_2$: C, 66.64; H, 7.74; N, 11.96. Found: C, 66.51; H, 7.81; N, 11.72.

***N*-Acetyl-*N'*-benzylethylenediamine.** A colorless solid of mp $55\text{--}57^\circ\text{C}$. IR (nujol): $\nu=3301, 3281, 1633\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.59$ (br s, 1H), 1.97 (s, 3H), 2.77 (t, $J=5.9$ Hz, 2H), 3.33 (dt, $J=11.5, 5.9$ Hz, 2H), 3.78 (s, 2H), 6.10 (br s, 1H), 7.23–7.37 (m, 5H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=23.37, 39.23, 48.03, 53.54, 126.94, 127.89, 128.30, 139.85, 169.93$. EI-MS: $m/z=193$ (MH^+), 192 (M^+), 133, 132, 120, 91 (bp). Anal. Calcd for $\text{C}_{11}\text{H}_{16}\text{N}_2\text{O}$: C, 68.72; H, 8.39; N, 14.57. Found: C, 68.43; H, 8.55; N, 14.43.

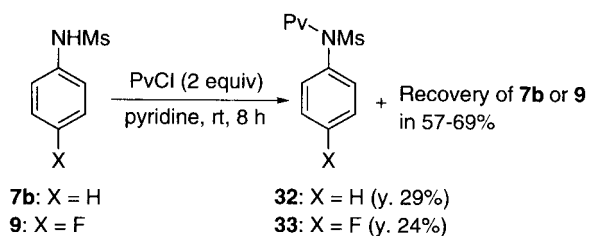
***N*-Benzyl-*N'*-benzoylethylenediamine.** Colorless plates (benzene–hexane) of mp $60\text{--}61^\circ\text{C}$. IR (nujol): $\nu=3352, 3309, 1631\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.97$ (s, 1H), 2.81 (dd, $J=5.9, 5.9$ Hz, 2H), 3.47 (dd, $J=5.9, 5.9$ Hz, 2H), 3.74 (s, 2H), 6.80 (br s, 1H), 7.15–7.45 (m, 8H), 7.65–7.74 (m, 2H). $^{13}\text{C NMR}$ (CDCl_3): $\delta=39.56, 47.88, 53.52, 126.77, 126.98, 127.95, 128.34, 131.16, 134.49, 139.93, 167.25$. EI-MS: $m/z=255$ (MH^+), 254 (M^+), 132 (bp), 106, 77. Anal. Calcd for $\text{C}_{16}\text{H}_{18}\text{N}_2\text{O}$: C, 75.56; H, 7.13; N, 11.01. Found: C, 75.78; H, 7.25; N, 10.99.

***N*-Benzoyl-*N'*-propylethylenediamine.** The physical data shown below were identical with those reported.^{3b} A pale yellow oil. IR (neat): $\nu=3300, 1644\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=0.90$ (t, $J=7.3$ Hz, 3H), 1.48 (tq, $J=7.3, 7.3$ Hz, 2H), 1.57 (br s, 1H), 2.58 (t, $J=7.3$ Hz, 2H), 2.83 (t, $J=5.9$ Hz, 2H), 3.49 (d, $J=5.9$ Hz, 1H), 3.52 (d, $J=5.9$ Hz, 1H), 6.98 (br s, 1H), 7.36–7.49 (m, 3H), 7.76–7.80 (m, 2H). FAB-MS: $m/z=207$ (MH^+).

***N*-Benzoyl-*N'*-propyl-1,3-propanediamine.** The physical data shown below were identical with those reported.^{3b} A pale yellow oil. IR (nujol): $\nu=3457, 3291, 1644\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=0.93$ (t, $J=7.3$ Hz, 3H), 1.55 (tq, $J=7.3, 7.3$ Hz, 2H), 1.57 (br s, 1H), 1.77 (tt, $J=5.9, 5.9$ Hz, 2H), 2.60 (t, $J=7.3$ Hz, 2H), 2.83 (t, $J=5.9$ Hz, 2H), 3.57 (br d, $J=5.9$ Hz, 1H), 3.59 (br d, $J=5.9$ Hz, 1H), 7.38–7.50 (m, 3H), 7.78–7.83 (m, 2H), 8.31 (br s, 1H). FAB-MS: $m/z=221$ (MH^+).

1-Benzoyl-3-methylpiperazine. The physical data shown below were identical with those reported.^{3b} A pale yellow oil. IR (nujol): $\nu=3308, 1623\text{ cm}^{-1}$. $^1\text{H NMR}$ ($\text{DMSO}-d_6, 100^\circ\text{C}$): $\delta=0.93$ (d, $J=6.2$ Hz, 3H), 2.09 (br s, 1H), 2.49–2.70 (m, 4H), 2.80–3.00 (m, 3H), 7.29–7.43 (m, 5H). FAB-MS: $m/z=205$ (MH^+).

***N*-Benzyl-*N'*-pivaloylethylenediamine.** A pale yellow oil. IR (neat): $\nu=3342, 1643\text{ cm}^{-1}$. $^1\text{H NMR}$ (CDCl_3): $\delta=1.19$



Scheme 2.

(s, 9H), 2.77 (t, $J=5.9$ Hz, 2H), 3.33 (dt, $J=5.9, 5.9$ Hz, 2H), 3.79 (s, 2H), 6.28 (br s, 1H), 7.21–7.37 (m, 5H). The signal of benzylic NH was not observed. ^{13}C NMR (CDCl_3): $\delta=27.67, 38.70, 39.03, 47.89, 53.40, 126.89, 127.87, 128.26, 139.96, 178.24$. EI-MS: $m/z=235$ (MH^+), 234 (M^+), 149, 133, 120, 91 (bp), 57. Anal. Calcd for $\text{C}_{14}\text{H}_{22}\text{N}_2\text{O}$: C, 71.76; H, 9.46; N, 11.95. Found: C, 72.02; H, 9.60; N, 11.84.

Acknowledgements

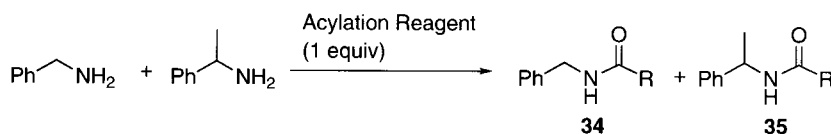
This study was financially supported by a Grant-in-Aid for Scientific Research from the Ministry of Education, Science and Culture, Japan, and the Sasakawa Scientific Research Grant from The Japan Science Society.

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- Preparation of **6** with BuLi and R^2COCl ($\text{R}^2\text{CO}=\text{Z}$, cyclohexylcarbonyl and pivaloyl) in THF were also unsuccessful.
- Benzyloxycarbonylation of *N*-phenyltrifluoromethanesulfonamide (**7a**) under ZCl in pyridine gave an *N*-benzylation product without formation of the desired *Z*-sulfonamide **12a**. A similar result has been reported in the benzyloxycarbonylation of *N*-(4-trifluoromethylphenyl)trifluoromethanesulfonamide by Yasuhara, Nagaoka and Tomioka: see, Ref. 2w.

Table 11. Competition acylation



Entry	Acylation reagent	Conditions	Yield (%) 34 + 35	Selectivity ^a 34 : 35
1	4a	THF, 0°C, 12 h	89	8:1
2	13a	THF, rt, 18 h	90	8:1
3	ZCl	Et_3N , THF, 0°C, 1 h	92	3:1
4	4d	THF, 0°C, 3 h	86	5:1
5	13d	THF, 0°C, 48 h	89	10:1
6	Bz_2O	THF, 0°C, 1 h	92	5:1
7	BzCN^b	CH_2Cl_2 , 0°C, 12 h	97	5:1
8	Bz_2NOMe^c	THF, 0°C, 12 h	88	3:1
9	4c	THF, 0°C, 10 h	86	9:1
10	Ac_2O	Et_3N , CH_2Cl_2 , 0°C, 1 h	91	1:1

^a Determined by ^1H NMR analysis.

^b Ref. 2k.

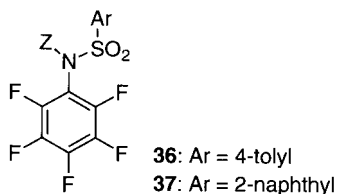
^c Ref. 2m.

8. *N*-acetyl-*N*-phenyltrifluoromethanesulfonamide has been reported to behave as an acetylation reagent: Hendrickson, J. B.; Bergeron, R. *Tetrahedron Lett.* **1973**, 4607–4610.

9. In this case, the use of other sulfonamides **7b** and **9** gave less satisfactory results (Scheme 2).

10. The *N*-*Z*-sulfonamide **27** acylated 2-(phenylethyl)amine (**17**) in 98% yield for 4 h at rt.

11. The selectivity in the acylation of a 1:1 mixture of a less hindered amine and a hindered amine was also investigated with **13a**, **13d**, **4a** and **4d**. The observed selectivity in the acylation with **13a**, **13d**, **4a** and **4d** was not at all inferior to the currently reagents (entries 3, 6–8 and 10), as shown in Table 11. The above competition acylation with the use of **36** and **37** instead of **4a** gave the same selectivity.



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